

# Stability Indicating UV-Visible Spectroscopic Method For Simultaneous Estimation Of Bismuth Sub-Citrate, Metronidazole And Tetracycline In Their Formulation

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ABSTRACT: A simple accurate, precise method was developed for the simultaneous estimation of the bismuth sub-citrate, tetracycline and metronidazole in their formulation. The method involved solving simultaneous equations based on measurement of absorbance at three wavelengths, 217nm, 249nm, and 289nm i.e.,  $\lambda$ max of the bismuth sub-citrate, tetracycline and metronidazole respectively. Beer's law was obeyed in the concentration range of 14-70µg/ml, 12.5-62.5µg/ml, and 1.60-4.80µg/ml for bismuth sub-citrate, tetracycline and metronidazole and correlation coefficient (r<sup>2</sup>) values are 0.9999, 0.9999 and 0.9998 respectively. The proposed method was validated as per the ICH guidelines for parameters like linearity, specificity, precision, accuracy, robustness and ruggedness. %RSD of the tetracycline, bismuth subcitrate and metronidazole were found to be 0.395, 0.232 and 0.601 respectively. The % mean recovery of bismuth sub-citrate, tetracycline and metronidazole was found to be 99.51%, 99.42% and 100.47% respectively. The method was accurate, specific, precise and found to be suitable for the quantitative analysis of the drug and dosage form. **KEYWORDS:**Simultaneous Estimation,

Metronidazole, Bismuth sub-citrate, Tetracycline, UV-visible spectroscopy<sup>1-6</sup>.

# I. INTRODUCTION

Simultaneous estimation method<sup>7</sup> (Vierodt's method) is used to determine the concentration of several components present in the same mixture even if their spectra overlap. If a sample contains two absorbing drugs (X and Y) each at which absorbs at the  $\lambda_{max}$  of the other, it may be possible to determine both drugs by technique of simultaneous equations that certain criteria apply. The information required is

- a) The absorptivities of X at  $\lambda_1$  and  $\lambda_2$ ,  $a_{x1}$  and  $a_{x2}$ , respectively
- b) The absorptivities of Y at  $\lambda_1$  and  $\lambda_2$ ,  $a_{y1}$  and  $a_{y2}$ , respectively
- c) The absorbances of the diluted sample at  $\lambda_1$  and  $\lambda_2$ ,  $A_1$  and  $A_2$  respectively.

Let  $C_x$  and  $C_y$  be the concentrations of X and Y respectively in the diluted sample.

Two equations are constructed based upon the fact that at  $\lambda_1$  and  $\lambda_2$  the absorbance of the mixture is the sum of the individual absorbances of X and Y.

At  $\lambda_1$ ,  $A_1 = a_{x1}bc_x + a_{y1}bc_y$ . (1) At  $\lambda_2$ ,  $A_2 = a_{x2}bc_x + a_{y2}bc_y$  (2)

At  $\lambda_2$ ,  $A_2 = a_{x2}bc_x + a_{y2}bc_y$ . (2) Then the concentrations can be calculated as

 $C_{x} = \frac{A_{2}ay_{1} - A_{1}ay_{2}}{ax_{2}ay_{1} - ax_{1}ay_{2}} \& C_{y} = \frac{A_{1}ax_{2} - A_{2}ax_{1}}{ax_{2}ay_{1} - ax_{1}ay_{2}}$ 

Bismuth sub-citrate<sup>15-17</sup> (bismuth; pentapotassium;2-hydroxypropane-1,2,3-

tricarboxylate) is a mineral and a soluble complex, bismuth salt of citric acid used to treat stomach ulcers caused by Helicobacter pylori infections in combination of metronidazole and tetracycline. Bismuth sub-citrate is very effective in the treatment of gastro duodenal disorders and seems to act via several mechanisms. It has little acid neutralizing effect and does effect acid secretion.It is uncertain whether it affects pepsin secretion, but it does inhibit peptic activity. It causes an increase in mucus glycoprotein secretion and may also bind to the gastric mucus layer to act as a diffusion barrier to HCl.It accelerates ulcer healing and causes an accumulation of epidermal growth factor around the ulcer. In addition, it has a cytoprotective effect and increases mucosal secretion of prostaglandins and bicarbonate. It quickens ulcer recuperating and



causes an amassing of epidermal development factor around the ulcer. It has bactericidal effects against Helicobacter pylori (which is associated with gastritis and peptic ulcers). It also prevents adhesion of H. pylori to epithelial cells and can inhibit enzymes secreted by H. pylori, such as protease, lipases, glycosidases and phospholipases.

Tetracyclines<sup>18</sup>((4S,4aS,5aS,6S,12aR)-4-(dimethylamino)-1,6,10,11,12a-pentahydroxy-6methyl-3,12-dioxo-4,4a,5,5a-tetrahydrotetracene-2carboxamide) are called broad – spectrum antibiotics because they can be utilized to treat a wide variety of bacterial infections like skin, eye, intestines, respiratory tract, urinary tract, lymph nodes, and other body systems. Doctors may prescribe these medications to treat pneumonia, gonorrhoea, Rocky Mountain spotted fever, and other infections caused by microorganisms. Tetracycline is a short-acting antibiotic that inhibits bacterial growth by inhibiting translation. It binds to the 30S ribosomal subunit and prevents the aminoacyl t-RNA from binding to the A site of the ribosome. It also binds to the 50S ribosomal subunit. This binding is reversible in nature. During protein biosynthesis, the new t-RNA with the amino acid attempts to bind to A-site of the ribosome.

Metronidazole<sup>19-21</sup>(1-(2-hydroxyethyl)-2methyl-5-nitroimidazole) is used to treat bacterial infections in different areas of the body. The extended release tablets are used to treat women with vaginal infections (bacterial vaginosis). It works by killing the bacteria or preventing their growth. Metronidazole is primarily used to treat pelvic inflammatory diseases (along with antibacterial like ceftriaxone), pseudomembranous colitis, aspiration pneumonia, rosacea(topical), fungating wounds(topical). intra-abdominal infections, lung abscess, periodontitis, amoebiasis and oral infections. It is also often used to eradicate Helicobacter pylori along with other drugs and to prevent infection in people recovering from surgery.



Structure of Bismuth sub-citrate



**Structure of Tetracycline** 





Structure of metronidazole

# II. MATERIALS AND METHODS

#### Instruments

A UV/VIS spectrophotometer, model LAB INDIA 3000<sup>+</sup>, with a spectral band width of 1.5nm with a pair of 10mm quartz cell was used for the experimental work.

#### **Reagents and chemicals**

All the chemicals used were of AR grade. Methanol and water was obtained from RANKEM, and the standards Bismuth sub-citrate, Tetracycline and Metronidazole was procured from KP LABS.

#### Procedures

#### **Preparation of standard solution**

Accurately weigh about 10mg of bismuth sub-citrate and transfer it into a 10ml volumetric flask add methanol and shake it. Final volume of solution was obtained by adding methanol up to the mark.  $(1000\mu g/ml)$  (Stock-I).

## **Preparation of sample solutions**

Pipette out 1ml from the standard solution and transfer it into a 10ml volumetric flask and make upto the mark with methanol (stock-II). From this  $10\mu g/ml$  concentrations were prepared and scanned in the entire UV range of 200-400nm.

# III. METHOD DEVELOPMENT

#### Wavelength selection

The Standard solutions of Metronidazole  $37.5\mu$ g/ml, Tetracycline  $37.5\mu$ g/ml and  $42\mu$ g/ml of Bismuth Sub citrate are scanned in UV Spectroscopy in the range of 200 nm to 400 nm.



Overlain spectra of Metronidazole, Tetracycline and Bismuth sub-citrate

#### Assay

The sample and standard solutions were scanned and the absorbances are taken.



DRUG	LABEL	%
	CLAIM	ASSAY
	(MG)	
METRONIDAZOLE	125	96.96
TETRACYCLINE	125	95.63
BISMUTH SUB	140	99.01
CITRATE		

Assay results of Metronidazole, Tetracycline and Bismuth sub-citrate

# **IV. VALIDATION PARAMETERS**

#### Linearity

The linearity of an analytical method is its ability to elicit test results that are directly proportional to the concentration of analytes in samples within a given range or proportional by means of well-defined mathematical transformations. Linearity was determined by series injections of whose concentrations span 50-150 of the expected concentration range. The linearity range was found to lie from  $12.5\mu$ g/ml to  $62.5\mu$ g/ml of Metronidazole, Tetracycline and 14 µg/ml to 70 µg/ml Bismuth Sub Citrate. The calibration curve of Metronidazole, Tetracycline and Bismuth sub-citrate are shown below.



#### Calibration curve of Metronidazole



**Calibration curve of Tetracycline** 





## Calibration curve of Bismuth sub-citrate

#### Accuracy

The accuracy of an analytical method is defined as the degree to which the determined value of analyte in a sample corresponds to the true value. The accuracy study was performed for 50%, 100% and 150 % for all drugs. Each level was canned intriplicate into the system. The Absorbance of each level was used for calculation of % recovery.

% CONCENT RATION (AT SPECIFIC LEVEL)	ABSORBAN CE	AMOUNT ADDED	AMOUNT FOUND	% RECOVERY	MEAN RECOVERY	
METRONIDA	ZOLE					
ACCURAC Y 50%	0.197	6.25	6.52	104.30		
ACCURAC Y 100%	0.377	12.5	12.18	97.47	100.47	
ACCURAC Y 150%	0.578	18.75	18.68	99.63		
TETRACYCLINE						
ACCURAC Y 50%	0.225	6.25	6.29	100.70		
ACCURAC Y 100%	0.446	12.5	12.23	97.83	99.42	
ACCURAC Y 150%	0.682	18.75	3.03	99.73		
BISMUTH SUB CITRATE						
ACCURAC Y 50%	0.376	7	7.01	99.93		
ACCURAC Y 100%	0.751	14	13.86	99.01	99.51	
ACCURAC Y 150%	1.133	21	20.91	99.58		



#### Precision

The precision of an analytical procedure expresses the closeness of agreement (degree of scatter) between a series of measurements obtained from multiple sampling of the same homogeneous sample under the prescribed conditions. Accurately weigh and transfer 10mg of formulation (3.205mg of Metronidazole, 3.205mg of Tetracycline and 3.58mg of Bismuth Sub-citrate) into a 10ml clean dry volumetric flask (STOCK-I) & add Diluent and sonicate to dissolve it completely and make volume up to the mark with the same solvent. Make

RESULTS	ABS OF METRONIDAZOLE AT 289 NM	ABS OF TETRACYCLINE AT 249 NM	ABS OF BISMUTH SUB-CITRATE AT 217 NM
ID Precision-1	0.388	0.458	0.754
ID Precision-2	0.386	0.454	0.752
ID Precision-3	0.384	0.452	0.753
ID Precision-4	0.387	0.453	0.756
ID Precision-5	0.385	0.454	0.754
ID Precision-6	0.385	0.452	0.757
AVERAGE	0.386	0.454	0.754
STANDARD DEVIATION	0.001	0.002	0.002
%RSD	0.382	0.491	0.247

#### Intermediate precision

To evaluate the intermediate precision (also known as Ruggedness) of the method, Precision was performed on different day.

Accurately weigh and transfer 10mg of formulation (3.205mg of Metronidazole, 3.205mg of Tetracycline and 3.58mg of Bismuth Sub-citrate) into a 10ml clean dry volumetric flask (STOCK-I) & add Diluent and sonicate to dissolve it completely and make volume up to the mark with the same solvent. Make further dilutions and Scan the solution 3 timess in range of 200-400nm.

	ABS OF	ABS OF	ABS OF BISMUTH		
RESULTS METRONIDAZOLE		TETRACYCLINE	SUB-CITRATE AT 217		
	AT 289 NM	AT 249 NM	NM		
Precision-1	0.385	0.452	0.755		
Precision-2	0.382	0.454	0.752		
Precision-3	0.386	0.451	0.754		
Precision-4	0.389	0.452	0.755		
Precision-5	0.384	0.456	0.753		
Precision-6	0.385	0.453	0.757		
AVERAGE	0.385	0.453	0.754		
STANDARD					
DEVIATION	0.002	0.002	0.002		
	0.002	0.002	0.002		
%RSD					
	0.601	0.395	0.232		



#### Robustness

Robustness tests examine the effect that operational parameters have on the analysis results. For the determination of a method's robustness, a number of method parameters, for example, pH, flow rate, column temperature, injection volume, detection wavelength or mobile phase composition, are varied within a realistic range, and the quantitative influence of the variables is determined. **SET-I**Accurately weigh and transfer 10mg of formulation (3.205mg of Metronidazole, 3.205mg of Tetracycline and 3.58mg of Bismuth Sub-citrate) into a 10ml clean dry volumetric flask (STOCK-I) & add Diluent and sonicate to dissolve it completely and make volume up to the mark with the same solvent. Make further dilutions and Scan the solution 3 times in range of 200-400nm.

SET-II repeat the same procedure and scan after 24hrs in the range of 200-400nm.

S.NO	WAVE LENGTH (NM)	ABSORBANCE		
ROBUSTNESS OF N	METRONIDAZOLE			
1	284 nm	0.388		
2	289nm	0.386		
3	294nm	0.387		
ROBUSTNESS OF T	ETRACYCLINE			
1	244nm	0.452		
2	249nm	0.455		
3	254nm	0.454		
ROBUSTNESS OF F	BISMUTH SUB-CITRATE			
1	212nm	0.755		
2	217nm	0.751		
3	222nm	0.752		

## V. DEGRADATION STUDIES

The International Conference on Harmonization (ICH) guideline entitled stability testing of new drug substances and products requires that stress testing be carried out to elucidate the inherent stability characteristics of the active substance. Stress degradation studies are carried out on the Metronidazole, Tetracycline and Bismuth sub-citrate using the proposed method.

## Forced degradation:

**Under acidic condition:**From the above prepared stock solution, pipette out 1ml and make up to 10ml with methanol in volumetric flask (stock-II). To this add 1ml of 0.1N HCl and keep it aside for an hour. Now the scan the solution in the range of 200-400nm.

**Under basic condition:** From the above prepared stock solution, pipette out 1ml and make up to 10ml with methanol in volumetric flask (stock-II). To this add 1ml of 0.1M NaOH and keep it aside for an hour. Now the scan the solution in the range of 200-400nm.

% DEGRA DATION RESULT S	METRONIDAZOLE		TETRACYCLINE		BISMUTH SUBCITRATE	
	ABS	% DEGRAD ATION	ABS	% DEGRADATION	ABS	% DEGRADATION
STAND ARD	0.386		0.455		0.757	
ACID	0.369	4.3	0.440	3.2	0.719	5.1
BASE	0.378	2.6	0.448	1.6	0.706	6.8



# **VI. CONCLUSION**

An attempt was made to develop and validate a stability indicating UV-VISIBLE method for the simultaneous estimation of the bismuth subcitrate, tetracycline and metronidazole in their formulation and the solvent used was methanol. The optimized wavelength for bismuth sub-citrate, tetracycline and metronidazole was at 281nm. The developed method is validated based on ICH guidelines.

The UV-Visible method was developed for the quantification of metronidazole, tetracycline and bismuth sub-citrate.

The linearity lies between 12.5-62.5  $\mu$ g/ml for metronidazole and tetracycline, 14 -70  $\mu$ g/ml for Bismuth sub-citrate.

The stress degradation studies was indicated for bismuth sub-citrate, tetracycline and metronidazole (below 10%), it shows no degradation was found in the proposed method.

The advantages present in the simplicity of sample preparation, the low costs of reagents used. The proposed method ensures sufficient and precise quantification of the compounds. Hence, the proposed method is rapid, accurate, precise, specific, robust, and economical. Results from statistical analysis of the experimental results were indicative of satisfactory precision and reproducibility. Hence, the proposed method can be used for routine drug analysis of Metronidazole, Tetracycline and Bismuth sub-citrate in their combination.

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